SYNTHESIS OF 6-(D-α-AMINOPHENYLACETAMIDO-1-14C)-PENICILLANIC ACID.

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SUMMARY

The synthesis of $6-(D-\alpha-aminophenylacetamido-1-^{14}C)$ -penicillanic acid (V) is described. Benzaldehyde (I) was converted with sodium cyanide- ^{14}C and ammonium chloride into D, $L-\alpha$ -aminophenylacetic acid- $1-^{14}C$ (II), from which the D-form was separated with D-camphorsulfonic acid. The $D-\alpha$ -aminophenylacetic acid- $1-^{14}C$ (III) was converted, with ethyl acetoacetate into enaminoderivative (IV), which, by means of reaction with ethyl chlorocarbonate and with triethylammonium salt of 6-aminopenicillanic acid, gave $6-(D-\alpha-aminophenylacetamido-1-^{14}C)$ -penicillanic acid(V).

INTRODUCTION

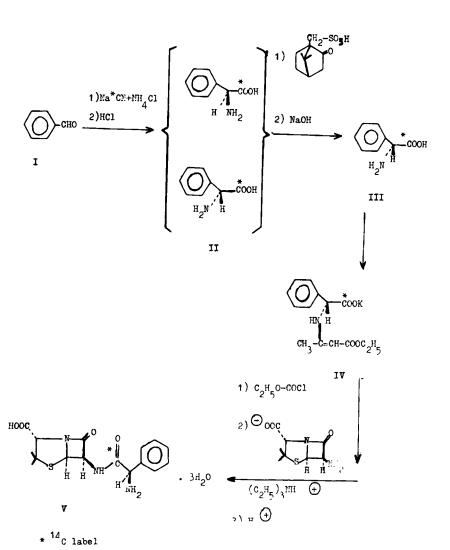
The $6-(D-\alpha-\text{aminophenylacetamido})$ -penicillanic acid (V), ampicillin, is one of the most useful antibiotics and one of most today used semisynthetic penicillins. It is not till now completely known the <u>in vivo</u> metabolism of ampicillin and for such purpose $6-(D-\alpha-\text{aminophenylacetamido}-1-\frac{1}{C})$ -penicillanic acid(V) was synthetized.

PESUITS

Benzaldehyde (I), by means of reaction with sodium cyanide 14 C and ammonium chloride and then by means of hydrolysis with hydrochloric acid according $^{(7)}$, was converted into 0 , $^{1-a}$ -aminophenylacetic acid-1- 14 C (II).

From 0,1-x-aminophenylacetic acid-1-10 (II), the D-form (III) was separated to D-camphorsulfonate. The 1-x-aminophenylacetic acid-1-10 remaining in the mother liming was again converted with 1a0H at 100° into D,L-form from which -x-amirophenylacetic acid (III) was again recovered, as indicated above, with -comphorsulforic acid.

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The compound (III) was treated with ethyl acetoacetate and potassium hydroxide and gave the enaminoderivative (IV).

The compound (IV), treated in acetone with ethyl chlorocarbonate, gave the mixed anhydride which was reacted with triethylamonium salt of 6-aminopenicillanic acid and gave $6-(D-\alpha-aminophenylacetamido-1-{}^{14}C)$ -penicillanic acid trihydrate (V).

EXPERIMENTAL

Materials and methods. -Melting points are uncorrected, optical rotation were determined at 20° with Perkin-Elmer mod. 141.

Infrared spectra were measured in nujol with Perkin-Elmer mod. 157 C. Chromatography:

10 mcg of D,L-a-aminophenylacetic acid-1-10 c or D-a-aminophenylacetic acid-1-10 c dissolved in 10 mcl of methanol, 20 mcg of potassium salt of N-(1-ethoxycarbo-nylpropen-2-yl)-D-a-aminophenylacetic acid-1-10 c in 10 mcl of HCl N in water, 100 mcg of ampicillin dissolved in 10 mcl of water, were used.

Thin layer chromatography method a: was performed on silica gel treated with buffer pH 6.0 (obtained from 135 ml of 0.1M Na₂HPO₄ and 865 ml of 0.1M KH₂PO₄) and desiccated 30 min at 105°, using as eluent the superior phase of n-butanol--acetic acid-water 4:1:5 V/V. After the elution and desiccation of the plate, it was sprayed with ninidrin 1% in ethanol and heated at 105° for 10 min.

Thin layer chromatography method b: was performed on Kieselgel GF 254 Merck using as eluent i-propanol-formic acid-water 40:2:10 V/V. After elution and designation the plate was sprayed with minidrin as before.

Thin layer chromatography method c: was obtained on Kieselgel GF 250 Merck using as eluent n-butanol-acetone-triethylamine-water 30:30:6:15 V/V and sprayin with minidrin as indicated before.

Thin layer chromatography method d: was obtained on Kieselgel FF 254 Merck treated with buffer pH 6.0 and desiccated 30 min at 1050, using as eluert i-propanol-formic acid-water 40:2:10 V/V and spraying with minidrin as indicated before.

Paper chromatography: was performed on Whatman n°4 using as eluent the superior phase of n-butanol-acetic acid-water 4:1:5 V/V. After 6 hours of descending

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chromatography, the paper was sprayed with NaOH 0.5N, heated 20 min at 50°, sprayed again with solution obtained from 24.5 g I₂, 8 g KI and 200 ml water and mixed with solution obtained from 1 g starch in 100 ml of water and 6 ml of acetic acid; white spots on dark background.

Chemical and biological analyses on the compound (V) were determined according C.F.R. of Food and Drug Administration and reported on anhydrous basis in comparison with an F.D.A. standard.

D, L-a-aminophenylacetic acid-1- C (II).

To a solution of 7.5 g of sodium cyanide (150 mmole) and 1 mCi of Amersham Na 14 CN in 40 ml of water were added 8.85 g of amonium chloride (165 mmole) and 15.9 g of benzaldehyde (150 mmole) dissolved in 40 ml of methanol. The solution was stirred 2 hours at 20-30°, diluted with 100 ml of water and extracted 2 x 50 ml of benzene. The benzene combined phases were washed 2 x 10 ml of water and then extracted 2 x 40 ml of HCl 20% in water. The acid extracts were refluxed 2 hours, diluted with 150 ml of water and evaporated in vacuo at 50° to 90 ml. The solution was treated with coal, filtered and treated with NH₄0H 32% W/V to pH 7. After 20 hours at 20° the solid product was filtered, washed with water, dissolved in 60-ml of N NaOH and 15 ml of methanol. The solution was again treated with coal, filtered and treated at 60° with HCl 20% in water to pH 6; after cooling at 20°, the D,L-α-aminophenylacetic acid-1- 14 C (II) was filtered, washed with water, dried at 50° (6.9 g), mp 265-8°, IR bands at 1610, 1550, 1510, 1200, 1130, 1070, 1030, 920, 900, 850, 750, 730, 690 cm 1.

Only one spot in thin layer chromatography (method a Rf 0.30, method b Rf 0.75, method c Rf 0.0, method d Rf 0.70). The product had SA 0.0364 µCi/mg in HCl N (a sample after 2 crystallizations from ethanol-water 1:1 to constant specific activity gave SA 0.0369 µCi/mg).

Chemical yield 30%, radiochemical yield 25.1%.

D-α-aminophenylacetic acid-1-16 (III).

D, L- α -aminophenylacetic acid-1- 14 C (II) (6.8 g, 45 mmole) in 30 ml of water was treated with 11.25 g of D-camphorsulfonic acid monohydrate (45 mmole) at 90 ° with stirring. The hot solution was filtered, cooled at 0° and the solid product was filtered, washed with 10 ml of water and dried giving 5 g of D-camphorsulfonate of D- α -aminophenylacetic acid-1- 14 C, $\lceil \alpha \rceil$ D -40° 2% in water. The

mother liquor was treated with NaOH 30% in water to pH 7 at 40°, the solid product was filtered, washed with water, dried giving 4.5 g of product which was refluxed 7 hours with 40 ml of NaOH 10% in water, treated with HCl 20% in water to pH 7, cooled at 0° and filtered giving 4 g of D,L-α-aminophenylacetic acid-1-\frac{14}{C} (II). From these 4 g of (II), by treatment with D-camphorsulfonic acid as indicated above, 3 g of D-camphorsulfonate of D-α-aminophenylacetic acid-1-\frac{14}{C} were recovered, \left(\alpha\right)_D -43° 2% in water. D-camphorsulfonate of D-α-aminophenylacetic acid-1-\frac{14}{C}(8 g) was dissolved in 30 ml of water at 80° and treated with NaOH 30% in water to pH 7. After 2 hours at 0° the solid product was filtered and dried in vacuo giving 2.9 g of (III), mp 247-51°, IR bands at 1610, 1550, 1510, 1200, 1130, 1070, 1030, 920, 900, 850, 750, 730, 690 cm \frac{1}{2}{2}{2}{2} \frac{148° 7%}{2} \text{in HCl N/2.}

Only one spot in the thin layer chromatography (method a Rf 0.35, method b Rf 0.80, method c Rf 0.0, method d Rf 0.70). The product had SA 0.0364 μ Ci/mg in NCl N(a sample after 3 crystallizations from ethanol-water 1:1 to constant specific activity gave SA 0.0353 μ Ci/mg).

Chemical yield 42.5%, radiochemical yield 42.5%.

Potassium salt of N-(1-ethoxycarbonylpropen-2-yl)-D-a-aminophenylacetic acid--1-14 C (IV).

To a solution of 1.27 g of KOH 85%(19.3 mmole) in 20 ml of methanol, 2.83 g of D- α -aminophenylacetic acid-1- 4 C (III) (18.7 mmole) were added; the mixture was treated at 60° with 2.68 g of ethyl acetoacetate (20.6 mmole) dissolved in 12 ml of methanol. The solution was refluxed 20 min, evaporated in vacuo to dryness and diluted with 40 ml of isopropanol at 20°. The solid product was filtered, washed with 10 ml of isopropanol and dried in vacuo giving 4.25 g of (IV), mp 214-6°, IR bands at 3300, 1650, 1610, 1570, 1270, 1170, 1130, 1070, 1060, 1030, 1010, 950, 940, 780, 730, 700, cm $^{-1}$, $\left[\alpha\right]_D + 121^\circ$ 5% in water.

Only one spot in the thin layer chromatography (method a Rf 0.40, method b of 0.90, method c Rf0.0, method d Rf 0.75). The product had SA 0.017 µCi/mg in water(a sample after 2 crystallizations from methanol-i-propanol 1:10 to constant specific activity gave SA 0.0175 µCi/mg).

Chemical yield 75%, radiochemical yield 70.1%

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6-(D-α-aminophenylacetamido-1-1C)-penicillanic acid trihydrate (V).

To 30 ml of acetone were added, at -40°, 4.2 g of (IV)(14 mmole) and then 1.63 g of ethyl chlorocarbonate (15 mmole), after 15 min at -40° 0.02 ml of N-methylmorpholine were added. The suspension was stirred at -40° for 30 min, and treated at -40° with a solution obtained from 3.24 g of 6-aminopenicillanic acid (15 mmole), 6.6 ml of water, 2.1 ml of triethylamine and 8.6 ml of acetone and previously cooled at -20°. The mixture was stirred 70 min at -40°, filtered, washed with 20 ml of acetone. The filtered solution was evaporated in vacuo at 20° to 20 ml, treated at 0° for 90 min with HCl 30% in water to pH 1, extracted 2 x 10 ml of chloroform. The aqueous phase was treated at 0° with NaOH 30% in water for 90 min to pH 4.5. After a night at 0°, the solid product was filtered, washed with water at 0°, dried at 40° in vacuo, giving 3.4 g of (V), humidity (Karl Fisher) 13.1%, $\{\alpha\}_D + 296^\circ$ 0.1% in water, iodometric assay 99% of the standard, bioassay 100% of the standard, amine titration 98% of the standard, acid titration 102% of the standard, IR bands at 3400, 1780, 1690, 1610, 1590, 1500, 1330, 1300, 1260, 1220, 1180, 1110, 700 cm⁻¹.

Only one spot in the thin layer chromatography with the same Rf as ampicillin standard(method a Rf 0.40, method b Rf 0.90, method c Rf 0.1, method d Rf 0.85) and only one spot in the paper chromatography with Rf 0.50 as the standard. SA 0.0121 μ Ci/mg in HCl N(a sample after 3 crystallizations from water gave SA 0.0125 μ Ci/mg).

Chemical yield 60%, radiochemical yield 57.6%.

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